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Seat No.:	Enrolment No.

GUJARAT TECHNOLOGICAL UNIVERSITY

BE - SEMESTER - VI EXAMINATION - WINTER 2015

Subject Code:162105 Date:10/12		2015	
•		Name: Electrometallurgy & Corrosion	
		60pm to 5:00pm Total Marks:	70
Instru		s: Attempt all questions.	
	2.	Make suitable assumptions wherever necessary. Figures to the right indicate full marks.	
Q.1	(a)	Define corrosion. Discuss about various factors affecting corrosion rate? Why corrosion should be prevented?	07
	(b)	Explain the following: 1. Faraday's laws 2. Anode 3. Cathode 4. Current Density	07
Q.2	(a)	Write Piling-Bedworth ratio. Discuss its importance and applications in corrosion study	07
	(b)	Discuss the Pourbaix diagram for Fe-H ₂ O system and show that how it is useful in corrosion study	07
	(b)	OR With proper examples explain how material selection is helpful to combat corrosion in automobile industries.	07
Q.3	(a)	Mention applications and limitations of Nernst's equation.	07
	(b)	When Intergranular corrosion occurs? Explain its mechanism and ways of prevention.	07
Q.3	(a)	OR What do you understand by pitting corrosion? Give the factors which increase pitting. Describe the different ways of prevention from	07
	(b)	pitting. What is galvanic series? Discuss its applicability and limitations in corrosion studies. Compare it with e.m.f. series.	07
Q.4	(a)	Discuss about concentration polarization. Differentiate between concentration polarization and activation polarization.	07
	(b)	Define Galvanizing. Explain Galvanizing Process. Give advantage and application of Galvanizing.	07
Q.4	(a)	OR What are inhibitors? Discuss different types of inhibitors with suitable example and their ways of functioning	07
	(b)	What is the principle of anodic and cathodic protection? Compare both techniques	07
Q.5	(a)	Describe anodizing process & explain the characteristics of Anodizing coatings.	07
	(b)	Discuss causes and mechanism of stress corrosion cracking. Suggest and explain possible remedial measures against it. OR	07
Q.5	(a)	Discuss mechanism of erosion corrosion. Mention the causes and way of combating erosion corrosion	07
	(b)	What do you mean by high temperature corrosion? Suggest suitable materials used for high temperature applications.	07