

**GUJARAT TECHNOLOGICAL UNIVERSITY****M.E. SEMESTER III-EXAMINATION – WINTER 2015****Subject code: 2732104****Date: 04/12/2015****Subject Name: Combustion Engineering****Time: 2:30 PM to 5:00 PM****Total Marks: 70****Instructions:**

1. Attempt all questions.
2. Make suitable assumptions wherever necessary.
3. Figures to the right indicate full marks.

- Q.1** (a) What is adiabatic flame temperature? Describe steady state analysis for open and closed system. **07**
- (b) Determine the stoichiometric equation for combustion of (a) Oxygen, (b) Air, (c) Excess fuel. **07**
- Q.2** (a) Determine the chemical kinematic reaction for (a) H<sub>2</sub>-O<sub>2</sub> system (b) Carbon monoxide oxidation. **07**
- (b) What is elementary reaction? Explain any two elementary reactions. **07**
- OR**
- (b) Consider a closed rigid vessel initially consists of H<sub>2</sub>. The reactor is rapidly heated to a high temperature T, and after heating and before decomposition the pressure becomes P<sub>0</sub>. The hydrogen then undergoes a unimolecular reaction
- $$\text{H}_2 \longrightarrow 2\text{H}$$
- During decomposition T, is maintained constant. Show that pressure history is given as  $P = [\text{H}_2]_0 \{ 2 - \exp(-k_{\text{uni}}(T) t) \} \bar{R}T$ .
- Q.3** (a) What is plug flow reactor? Derive expression for it. **07**
- (b) Explain constant volume and constant pressure reactor for a fixed mass. **07**
- OR**
- Q.3** (a) Consider the reaction of CO with O<sub>2</sub>, where air serves as source of O<sub>2</sub>: **07**
- (a) Show that  $\sum \dot{w}_k''' = 0$ .
- (b) Write the mass and species conservation equations for a 2-D flow system. Treat N<sub>2</sub> as inert gas.
- (c) If the density is maintained constant, simplify the mass conservation relation.
- (b) What is flame stabilization? Explain the stability of flame in premixed gas burner. **07**
- Q.4** (a) Show that  $I_f$  of a cylindrical jet is an invariant. What is the physical meaning of  $I_f$ ? **07**
- (b) Explain laminar jet with its physical description and chemical reaction. **07**
- OR**
- Q.4** (a) Describe the droplet evaporation for diffusion flame. Also find out its governing and conservation equations. **07**
- (b) Define the turbulent flow. Determine the governing equations for axisymmetric turbulent flow. **07**
- Q.5** (a) Under diffusion-controlled combustion, a spherical carbon particle producing CO will burn faster compared with a similar carbon sphere burning to CO<sub>2</sub>. True or false? Explain. **07**
- (b) Explain the modeling approach for premixed and partially premixed turbulent flames. **07**
- OR**
- Q.5** (a) Discuss the characteristic scale of wrinkles in turbulent premixed flame. **07**

(b) Derive the transport equation for  $\tilde{\xi}^{n^2}$  from equation

07

$$\frac{\partial}{\partial t}(\rho \xi) + \frac{\partial}{\partial x_k}(\rho u_k \xi) = \frac{\partial}{\partial x_k}(\rho D \frac{\partial \xi}{\partial x_k})$$

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